Numerical treatment of ion collisions with water molecules

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Synopsis A lattice treatment has been implemented to study collisions of multiply charged ions with H₂O within the independent electron approximation. The method has been applied to calculate ionization and electron capture probabilities for collisions of H⁺, He²⁺, Li³⁺ and C⁶⁺ with H₂O, at energies $20 \le E \le 300 \text{ keV/u}$.

In a previous work[1], we applied a lattice method to solve the time-dependent Schrödinger equation (TDSE) that appears in the semiclassical description of ion-atom collisions:

$$\widehat{H} - i\frac{d}{dt}\Psi(\vec{r},t) = 0, \qquad (1)$$

Ĥ where isthe clamped-nuclei Born-Oppenheimer electronic hamiltonian of the quasimolecule formed during the collision and Ψ is the electronic wavefunction. In that work we employed a modified version of the code GridTDSE code [2], originally designed to study the time evolution of nuclear wavepacket. In the lattice method the wavepacket is represented by a vector $\vec{\Psi}$ that stores the values of Ψ on the points of a 3D cartesian grid at each value of t. The generalization to ion-molecule collisions is straightforward at high collision energies, by employing the Franck-Condon approximation that assumes that the nuclei of the molecular target remain fixed at their equilibrium positions, \vec{R}_e , during the collision. The generalization to manyelectron systems involves the use of the independent electron approximation where each electron moves in an effective potential $\widehat{V}_{ef}(\vec{r}_i; \vec{R}_e)$ that includes the electron interaction with the core formed by the nuclei and the other electrons and $\widehat{H} = \widehat{T} + \widehat{V}_{ef} - Z/r_Z$, where Z is the charge of the ion.

The lattice method was applied to study electron-loss in $H^+ + H_2O$ collisions in Ref. [3].In the present work we consider the application of this method to collisions of He^{2+} , Li^{3+} and C^{6+} with H_2O . We have also calculated the electron capture probabilities as explained in Ref. [1].

The application of our numerical method re-

quires large computing resources, but it is very useful to support close-coupling and classicaltrajectory-Monte-Carlo calculations. It also allows us to follow the time evolution of the electronic wavepacket, as illustrated in Fig. 1 for electron capture in $\text{Li}^{3+} + \text{H}_2\text{O}$ collisions. The snapshots of the figure have been taken at the point of closest approach Y = 0. and at $Y = 27.6 \text{ a}_0$, where the molecule has left the box.



Figure 1. Contour plots of the electron density on the XY plane for the collisions $\text{Li}^{3+} + \text{H}_2\text{O}$ (1b₂, 1b₁). In the calculation the Li^{3+} is fixed a the origin and the molecule follows a rectilinear trajectory $\vec{R} = \vec{b} + \vec{v}t$ with v = 1.8 a.u, b = 2.0 a₀, $\vec{v} \parallel \hat{\vec{Y}}$ and $\vec{b} = b\hat{\vec{X}}$. During the collision the molecular plane is parallel to the XZ plane and the snapshots have been taken at Y = 0. (bottom panels) and Y = 27.6 a₀. (top panels).

This work has been partially supported by Ministerio de Ciencia, Innovación y Universidades (Spain) (project FIS2017-84684-R)

References

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